



Artificial neural networks analysis used to evaluate the molecular interactions between selected drugs and human α_1 -acid glycoprotein

Adam Buciński^{a,*}, Małgorzata Wnuk^a, Krzysztof Goryński^a, Anna Giza^a,
Joanna Kochańczyk^a, Alicja Nowaczyk^b, Tomasz Bączek^{c,d}, Antoni Nasal^c

^a Department of Biopharmacy, Faculty of Pharmacy, Collegium Medicum, Nicolaus Copernicus University, M. Skłodowskiej-Curie 9, 85-094 Bydgoszcz, Poland

^b Department of Organic Chemistry, Faculty of Pharmacy, Collegium Medicum, Nicolaus Copernicus University, M. Skłodowskiej-Curie 9, 85-094 Bydgoszcz, Poland

^c Department of Biopharmaceutics and Pharmacodynamics, Medical University of Gdańsk, Gen. J. Hallera 107, 80-416 Gdańsk, Poland

^d Department of Medicinal Chemistry, Faculty of Pharmacy, Collegium Medicum, Nicolaus Copernicus University, M. Skłodowskiej-Curie 9, 85-094 Bydgoszcz, Poland

ARTICLE INFO

Article history:

Received 15 July 2008

Received in revised form 5 November 2008

Accepted 6 November 2008

Available online 17 November 2008

Keywords:

ANN analysis

QSRR

α_1 -Acid glycoprotein (AGP) column

Molecular descriptors

ABSTRACT

Quantitative structure–retention relationships (QSRR) were proposed for α_1 -acid glycoprotein (AGP) column using physicochemical molecular descriptors of the selected drugs and interacting with that column. The set of 52 structurally diverse drug compounds, with experimentally derived logarithms of retention factors ($\log k$) values was considered. Thirty-six physicochemical property descriptors were calculated by standard molecular modeling and used to establish QSRR and predict the retention data by artificial neural network (ANN). The QSRR indicated that heat of formation (HF), Moriguchi *n*-octanol–water partition coefficient ($M \log P$) and the energy of the highest occupied molecular orbital (HOMO) are the most important for interactions between drugs and AGP. The proposed ANN model based on selected molecular descriptors showed a high degree of correlation between $\log k$ observed and computed. The final model possessed a 36-5-1 architecture and correlation coefficients for learning, validating and testing sets equaled 0.975, 0.950 and 0.972, respectively.

© 2008 Elsevier B.V. All rights reserved.

1. Introduction

Affinity chromatography (AC) is a liquid chromatography technique based on reversible interactions between the binding site of a macromolecule and an analyte molecule. Affinity chromatographic systems are obtained by immobilization of one of the pair of interacting molecules on a solid support and packing it into a column [1]. The stationary phase in AC is the main factor controlling the separation of compounds. Protein stationary phases were introduced first in the early 1980s [2–4]. For enantiospecific separations on AC supports containing ovomucoid [5], flavoprotein [6], avidin [7] and pepsin [8] were developed. Macromolecules currently used to form AC stationary phases are: human serum albumin [9], α_1 -acid glycoprotein (AGP) [10], keratin [11], collagen [12], melanin [13], amylose tris(3,5-dimethylphenylcarbamate) [14] and the basic fatty acid-binding protein from chicken liver [15].

Affinity chromatography, followed by quantitative structure–retention relationships (QSRR) analysis, provides information on both the analytes and the macromolecules forming the stationary phases. QSRR equations derived for test series of analytes (often drugs) are interpreted in terms of structural requirements

of the specific binding sites on macromolecules. Chromatographically demonstrated differences in analyte/macromolecule interactions may be relevant in view of molecular pharmacology and rational drug design. Moreover, specific high-performance affinity-chromatographic separations can be optimized by rational selection of chiral columns, the characteristics of which are provided by QSRR.

Barbato et al. [16] derived the relationships between the retention on AGP column and the lipophilic parameters of 23 amines. It was concluded that only the (*S*)-forms of neutral congeners get into mainly lipophilicity-driven interactions with AGP. On the other hand, the (*R*)-forms interact by a more complex mechanism, not exactly explained by $\log P$ or chromatographically determined lipophilicity parameter obtained on the so-called immobilized artificial membrane (IAM) column ($\log k_{\text{wIAM}}$).

An AGP column was employed in the experiments performed by Kaliszan et al. [17]. The aim of that study was to characterize structurally the binding site for organic-base drugs on the protein stationary phase. For a short series of β -adrenolytic drugs, for which the AGP-binding data determined by a standard biochemical procedure were available, a good correlation was found between the percent binding and $\log k$ from AC. The same authors [18] examined the retention mechanism on an AGP column of 16 antihistamine drugs. It appeared that the $\log k_{\text{AGP}}$ values from AC correlated significantly with the chromatographic hydrophobicity

* Corresponding author. Tel.: +48 52 585 39 09.

E-mail address: kizbiofarmacji@cm.umk.pl (A. Buciński).

parameter determined on IAM column ($\log k_{IAM}$). In a detailed QSRR analysis of $\log k_{AGP}$ data, the structural parameters reflecting the molecular size of the analyte (S_T), and the electron excess charge on the aliphatic nitrogen (N_{ch}), also appeared statistically significant.

A further investigation on an AGP column included a wider group of analytes [18]. Retention data ($\log k_{AGP}$), were determined for 52 basic drugs of diverse chemical structures and pharmacological activities. Among them, one could find: antagonists of histamine H1 and H2 receptors, antagonists of β -adrenoceptors, and drugs acting on α -adrenergic receptors.

Accurate predictions of retention could be achieved theoretically, if the nature of intermolecular interactions determining molecular recognition of the analytes by the counterparts forming the chromatographic systems were properly quantified. But that situation is rather unrealistic. Therefore, in chemical practice approximate, predictions however useful, can be realized which are valid in statistical terms rather than in strict thermodynamic categories [19]. QSRR are statistically derived relationships between dependent variable (a chromatographic parameter) and independent variables (the descriptors characterizing the molecular structure of analytes). QSRR have been applied not only for evaluate properties of HPLC stationary phases (e.g., to predict relative differences in binding activity of drugs to AGP immobilized on the silica surface), but also to: (i) predict retention for a new analyte, (ii) get insight into the molecular mechanism of separation operating in a given chromatographic system, (iii) identify the most informative structural descriptors of analytes and (iv) evaluate complex physicochemical properties of analytes [20].

In the chemical-property-prediction studies a few standard calculation procedures are employed. QSRR are most commonly derived by multiple regression analysis (MRA) [20]. The fundamental problem with multiple regression is that considering simultaneously a number of structural parameters (independent variables) cannot be mutually related, i.e., they should be as much orthogonal as possible. At the same time, the properties within the series of analytes requested to derive statistically significant and physically meaningful QSRR should be evenly distributed and cover a wide range of individual structural descriptor values. In addition, the series of model analytes must be large enough to exclude chance correlations but not too big to save time and effort necessary for chromatographic and structural analysis [20]. It may appear that for individual series of analytes it is impossible to observe all those requirements. The question arises then if other data processing methods are able to provide acceptable retention prediction. Specially promising from that point of view appear currently the artificial neural networks (ANN).

The artificial neural network analysis is a method of data analysis, which is to emulate the human brain's way of working. Neural nets exhibit the way in which arrays of neurons probably function in biological learning and memory. ANN differs from classical computer programs in that they "learn" from a set of examples rather than are programmed to get the right answer. The information is encoded in the strength of the network's "synaptic" connections [21]. In chemistry and related fields of research a consequently increasing interest in neural-network computing has been noted since 1986. Very recently several attempts were reported to use ANN to model chromatographic retention [22]. ANNs found also application to compound classification, modeling of structure–activity relationships [23], identification of potential drug targets and the localization of structural and functional features of biopolymers [24].

The aim of the current study was to design and test the appropriate ANN, which could allow to predict chromatographic retention on the basis of structural descriptors describing the structure of the selected basic drugs.

2. Materials and methods

2.1. Structural parameters from molecular modeling

Descriptors of the structure of drugs were calculated by standard molecular modeling. HyperChem program for personal computers with the extension ChemPlus (Hypercube, Waterloo, Canada) was used. The software performed geometry optimization by the molecular mechanics MM+ force field method which was followed by quantum chemical calculations according to the semi-empirical AM1 method. Moreover, the set of structural descriptors was supplemented with Dragon software (Milan Chemometris and QSAR Research Group, Milan, Italy). The list of descriptors is presented in Table 1.

2.2. RP HPLC retention data of drugs

A Merck-Hitachi (Vienna, Austria) HPLC system was employed for chromatographic measurements of binding of the compounds

Table 1
List of structural parameters of drugs employed in ANN analysis.

Number	Name	Descriptor
Electronic parameters		
1.	Dipole moment	μ
2.	HOMO energy	HOMO
3.	LUMO energy	LUMO
4.	Energy difference between molecular orbitals (LUMO and HOMO)	DLH
5.	Dielectric energy	DE
Parameters reflecting the size (bulkiness) of the agents		
6.	Atom count	AC
7.	Molecular weight	MW
8.	Molar refractivity	MR
9.	Molar refractivity-GC	MR-GC
10.	Molecular connectivity index of zero order	X-0
11.	Molecular connectivity index of first order	X-1
12.	Molecular connectivity index of second order	X-2
13.	Valence connectivity index of zero order	X0vC
14.	Valence connectivity index of first order	X1vC
15.	Valence connectivity index of second order	X2vC
16.	Molecular shape index of first order	κ -1
17.	Molecular shape index of second order	κ -2
18.	Molecular shape index of third order	κ -3
19.	Conformation minimum energy	CME
20.	Steric energy	SE
21.	Sum of atomic Van der Waals volumes (scaled on a carbon atom)	Sv
22.	V total size index/unweighted	Vu
23.	V total size index/weighted by atomic masse	Vm
24.	V total size index/weighted by atomic van der Waals volumes	Vv
25.	V total size index/weighted by atomic Sanderson electronegatives	Ve
26.	V total size index/weighted by atomic polarizabilities	Vp
27.	V total size index/weighted by atomic electrotopological states	Vs
28.	Sum of Kier–Hall electrotopological states	Ss
29.	Fragment-based polar surface area	PSA
30.	Solvent accessibility surface area	SASA
31.	Polarizability	P
32.	Sum of atomic Sanderson electronegatives (scaled on a carbon atom)	Se
33.	Sum of atomic polarizabilities (scaled on a carbon atom)	Sp
34.	Total energy	TE
35.	Heat of formation	HF
Lipophilicity parameters		
36.	Moriguchi <i>n</i> -octanol–water partition coefficient	M log P
Logarithm of HPLC retention factor (AGP column) – experimental		log <i>k</i> (AGP)

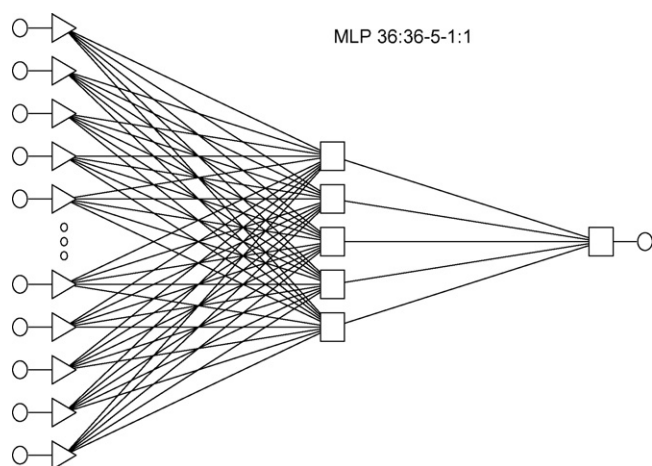


Fig. 1. Architecture of artificial neural network predicting chromatographic retention on the basis of selected structural descriptors. ANN model type: MLP 36:36-5-1:1.

studied to AGP [18]. The retention factors of the compounds studied were determined isothermally on Chiral AGP column 100 mm × 4 mm I.D. (ChromTech, Norsborg, Sweden) packed with α_1 -acid glycoprotein chemically bound to silica particles of 5 μ m diameter. The mobile phase was isopropanol – 0.01 M Sørensen phosphate buffer pH 6.5 (5:95, v/v). The mobile phase flow-rate was 0.5 ml/min. The detection wavelength was 215 nm. Logarithms of retention factors, $\log k_{\text{ACP}}$, were calculated taking the sodium nitrate peak as a measure of dead volume. A 1 mg amount of a drug solute was dissolved in 4 ml of methanol. The solution was diluted 10-fold with methanol and 20 μ l of the final solution was injected onto the column. The logarithms of retention factors of a series of drugs determined on AGP column taken from literature [18] were collected in Table 2.

2.3. Artificial neural network (ANN) analysis

HPLC retention data on AGP column – $\log k$ (AGP), for all the analytes from Table 1 were divided randomly into three groups. Variables for the analyzed drugs were divided into learning set with 26 compounds, validation set with 16 compounds and testing set with 10 compounds. Fig. 1 presents the architecture of the ANN model used for predictions of molecular interactions between AGP and selected drugs. An artificial neural network based on a multilayer perceptron consisting of 36 artificial neurons in the input layer, five in the hidden layer and one neuron in the output layer was used. A two-stage procedure with back-propagation and conjugate gradient descent methods were used to train the network. In the case of the network applied, learning was completed in 100 epochs by back-propagation method and 37 epochs by conjugate gradient descent method. First, ANN analysis was performed with the training and validation sets of data by means of iterative minimalization procedure allowing to optimize parameters of the network. Data from the learning set were presented in a randomized manner during the learning process. The third data set (test set) was served as checking of the generalization ability of the learned ANN.

3. Results and discussion

The list of numerical values of the structural parameters of the drugs studied derived from calculation chemistry, reflecting their electronic properties, size (bulkiness) and lipophilicity are summarized in Table 1. The final model possessed a 36-5-1 architecture and correlation coefficients for learning, validating and testing sets equaled 0.975, 0.950 and 0.972, respectively (Table 3).

Table 3
Statistics of ANN processing used during the study.

Statistics	Learning set	Validating set	Testing set
Mean	0.951	1.111	1.108
Data S.D. ^a	0.379	0.414	0.540
Error Mean ^b	-0.006	0.024	0.037
Error S.D. ^c	0.084	0.130	0.173
Abs. E. Mean ^d	0.067	0.107	0.144
Correlation ^e	0.975	0.950	0.972

^a Standard deviation of the target output variable.

^b Average error of the output variable.

^c Standard deviation of errors for the output variable.

^d Average absolute error (difference between target and actual output values) of the output variable.

^e The standard Pearson-R correlation coefficient between the target and actual output values.

An ANN model was used to correlate chromatographic behavior of the set of structurally diverse drugs with their structural descriptors and to create a model useful to prediction of retention values. A correlation between experimental and predicted $\log k$ (AGP) values in learning, validating and testing set is given in Fig. 2.

Table 1 contains the list of the structural parameters of the drugs studied derived from calculation chemistry, reflecting their electronic properties, size (bulkiness) and lipophilicity. The numerical values of descriptors along with $\log k$ (AGP) of the agents are summarized in Table 2. In Table 4 the results of sensitivity analysis of inputs are presented, which one was used to identify significance of individual molecular descriptors and to select descriptors that were considered the most important.

Using the proposed method, i.e., artificial neural network, it was possible to predict what physicochemical property descriptors influence on interactions between AGP and selected drugs (the sensitivity above one). Molecular descriptors with sensitivities lower than one were seemed to be detrimental to the model ANN.

It is rather expected that highly significant structural parameters for ANN processing of retention data are lipophilicity of drugs (CM $\log P$) and their electronic descriptors (highest occupied molecular orbital (HOMO) and DLH). They correspond to second, fourth and eleventh sensitivity ranks, respectively. The results obtained confirmed also that the bulkiness of the molecules of basic drugs is also very important for their binding to AGP. This physicochemical property is reflected mainly by the following parameters: heat of formation (HF), AC, MR, as well as by connectivity indices: X-1, κ -3, X1vC. The first order connectivity index, X-1 encodes single bond properties and κ present information concern-

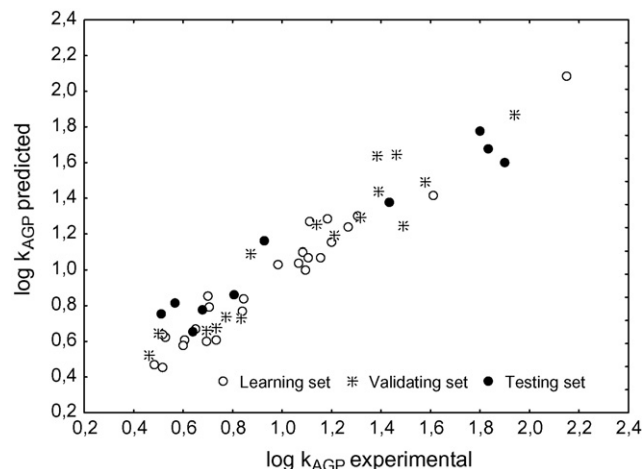


Fig. 2. Correlations between the calculated and the experimental retention data determined on an α_1 -acid glycoprotein column.

Table 4
Sensitivity analysis results for the structural parameters of drugs considered in ANN analysis.

Descriptor	Error	Rank
HF	1.620	1
M log P	1.478	2
X-1	1.241	3
HOMO	1.185	4
AC	1.166	5
MR	1.162	6
κ -3	1.160	7
X1vC	1.134	8
P	1.127	9
Sp	1.104	10
DLH	1.104	11
CME	1.103	12
Se	1.067	13
PSA	1.059	14
κ -2	1.059	15
X2vC	1.058	16
Ss	1.048	17
Vu	1.031	18
SASA	1.031	19
Ve	1.030	20
Sv	1.022	21
X-2	1.014	22
X-0	1.013	23
TE	1.010	24
Vs	1.006	25
μ	0.998	26
Vm	0.996	27
LUMO	0.994	28
X0vC	0.990	29
κ -1	0.990	30
MW	0.981	31
Vv	0.969	32
DE	0.968	33
Vp	0.964	34
MR-GC	0.961	35
SE	0.955	36

ing the shape, size, branching pattern and similarity of molecular graphs.

In fact, it was rather surprising that heat of formation occurred the most significant descriptor affecting the retention of the agents on AGP column. Heat of formation reflect basically the differences in bulkiness among the analytes [20], hence that descriptor can be treated as describing significantly a geometry of the molecular structure. The other significance values for sensitivity ranks were polarizability parameters (*P* and *Sp*), conformation minimum energy (CME), polar surface area (PSA) and further connectivity and topological shape indices (connectivity indices X-1 and κ -3) (Table 4). PSA can provide information about surface diffusion, absorption, contact surface and information about size of the molecules. In turn the contact surface area can be used an accurate predictor of water solubility and can be viewed as advice of the

extent to which the solute is exposed to intermolecular interaction with the solvent. PSA can also be useful tool to indicates the possibility of a compound to form hydrogen bonds which are an essential component of intermolecular interaction, e.g., protein–drug.

4. Conclusions

In the present study, a set of thirty-six descriptors, including both stationary phase and analytes properties, is adopted to build a QSRR model able to describe the retention behaviour of 52 basic drugs of diverse chemical structures as follows antagonists of histamine H1 and H2 receptors, β -adrenolytics, and drugs acting on α -adrenoreceptors. A artificial neural network provides an accurate QSRR model. In addition, ANNs model is able to detect relationships between depend ($\log k$) and independent (descriptors) variables. Finally, AGP columns can serve as instrument with ability to demonstrate types of interactions between acid glycoprotein and drugs. Useful information can be derived from the chemical structure of a drug to calculate descriptors describing various properties of that drug.

References

- [1] D. Hage, in: L.E. Schoeff, R.H. Williams (Eds.), Principles of Laboratory Instruments, Mosby-Year Book, St. Louis, 1993, pp. 196–197.
- [2] N.D. Danielson, R.W. Siergiej, Biotechnol. Bioeng. 23 (1981) 1913–1917.
- [3] S. Kamada, M. Maeda, A. Tsiu, Y. Umezawa, T. Kurahashi, J. Chromatogr. 239 (1982) 773–783.
- [4] S.C. Crowley, R.R. Walters, J. Chromatogr. 266 (1983) 157–162.
- [5] T. Miwa, M. Ichikawa, M. Tsuno, T. Hattori, T. Miyakawa, M. Kano, Y. Miyake, Chem. Pharm. Bull. 35 (1987) 682–686.
- [6] N. Mano, Y. Oda, N. Asakawa, Y. Yoshida, T. Sato, T. Miwa, J. Chromatogr. 623 (1992) 221–228.
- [7] J. Haginaka, T. Murashima, C. Seyama, J. Chromatogr. A 677 (1994) 229–237.
- [8] B. Sebillie, R. Zini, C.V. Madjar, N. Thuaud, J.P. Tillement, J. Chromatogr. 531 (1990) 51–77.
- [9] E. Domenici, C. Bertucci, P. Salvadori, G. Felix, I. Cahagne, S. Motellier, I.W. Wainer, Chromatographia 29 (1990) 170–176.
- [10] J. Hermansson, J. Chromatogr. 269 (1983) 71–80.
- [11] M. Turowski, R. Kaliszczan, J. Pharm. Biomed. Anal. 15 (1997) 1325–1333.
- [12] M. Turowski, R. Kaliszczan, Biomed. Chromatogr. 12 (1998) 187–192.
- [13] R. Kaliszczan, A. Kaliszczan, I.W. Wainer, J. Chromatogr. 615 (1993) 281–288.
- [14] T. Booth, I.W. Wainer, J. Chromatogr. A 737 (1996) 157–169.
- [15] G. Massolini, E. De Lorenzi, E. Calleri, C. Bertucci, H.L. Monaco, M. Perduca, G. Caccialanza, I.W. Wainer, J. Chromatogr. B 751 (2001) 117–130.
- [16] F. Barbato, F. Quaglia, M.T. Quercia, M.I. La Rotonda, Helvetica Chim. Acta 83 (2000) 767–776.
- [17] R. Kaliszczan, A. Nasal, M. Turowski, J. Chromatogr. A 722 (1996) 25–32.
- [18] R. Kaliszczan, A. Nasal, M. Turowski, Biomed. Chromatogr. 9 (1995) 211–215.
- [19] R. Kaliszczan, Chem. Rev. 107 (2007) 3212–3246.
- [20] R. Kaliszczan, Structure and Retention in Chromatography. A Chemometric Approach, Harwood Academic Publishers, Amsterdam, 1997.
- [21] J. Zupan, J. Gasteiger, Neural Networks for Chemists. An Introduction, VCH, Weinheim, 1993.
- [22] J. Zupan, J. Gasteiger, Anal. Chim. Acta 248 (1991) 1–30.
- [23] A. Nasal, A. Wojdełko, T. Bączek, R. Kaliszczan, M. Cybulski, Z. Chilmoneczyk, J. Sep. Sci. 25 (2002) 273–279.
- [24] A. Ajay, J. Med. Chem. 36 (1993) 3565–3571.